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# JEE Advanced: Paper-2 (2015)

# **IMPORTANT INSTRUCTIONS**

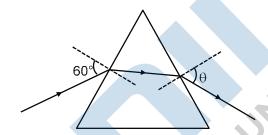
- The question paper has three parts: Physics, Chemistry and Mathematics. Each part has three sections.
- 2. Section 1 contains 8 questions. The answer to each question is a single digit integer ranging from 0 to 9 (both inclusive).
  - Marking Scheme: +4 for correct answer and 0 in all other cases.
- Section 2 contains 8 multiple choice questions with one or more than one correct option.
   Marking Scheme: +4 for correct answer, 0 if not attempted and –2 in all other cases.
- 4. Section 3 contains 2 "paragraph" type questions. Each paragraph describes an experiment, a situation or a problem. Two multiple choice questions will be asked based on this paragraph. One or more than one option can be correct.

# PART A: PHYSICS

#### **SECTION 1**

SECTION 1 (Maximum Marks: 32)

- This section contains EIGHT questions
- The answer to each question is a SINGLE DIGIT INTEGER ranging from 0 to 9, both inclusive
- For each question, darken the bubble corresponding to the correct integer in the ORS
- Marking scheme:
  - If the bubble corresponding to the answer is darkened
  - In all other cases
- 1. A monochromatic beam of light is incident at 60° on one face of an equilateral prism of refractive index n and emerges from the opposite face making an angle  $\theta(n)$  with the normal (see the figure). For  $n=\sqrt{3}$ the value of  $\theta$  is  $60^{\circ}$  and  $~\frac{d\theta}{dn}=~m$  . The value of m is



Ans. [2]

**Sol.** 
$$\sin 60^{\circ} = n \sin r_1$$

$$n \sin r_2 = \sin Q$$

$$n \sin (60^{\circ} - r_{\downarrow}) = \sin Q$$

n (
$$\sin 60^{\circ} \cos r_1 - \cos 60^{\circ} \sin r_1$$
) =  $\sin Q$ 

n 
$$\sin r_2 = \sin Q$$
  
n  $\sin (60^\circ - r_1) = \sin Q$   
n  $(\sin 60^\circ \cos r_1 - \cos 60^\circ \sin r_1) = \sin Q$   

$$\frac{\sqrt{3}n}{2} \cos r_1 - \frac{n}{2} \sin r_1 = \sin Q$$

$$\frac{\sqrt{3}n}{2}\sqrt{1-\frac{3}{4n^2}}-\frac{1}{2}\frac{\sqrt{3}}{2}=\sin Q$$

$$\frac{\sqrt{3}}{2} \frac{n}{2n} \sqrt{4n^2 - 3} - \sqrt{4n^2 - 3} - \frac{\sqrt{3}}{4} = \sin Q$$

$$\frac{4}{\sqrt{3}} \sin Q = \sqrt{4n^2 - 3} - 1$$

On differentiation

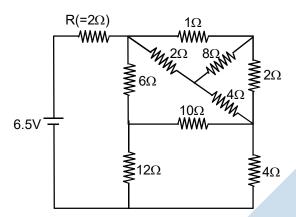
$$\frac{4}{\sqrt{3}} \cos Q \frac{dQ}{dn} = \frac{1}{2\sqrt{4n^2 - 3}} 8n$$

Putting the values

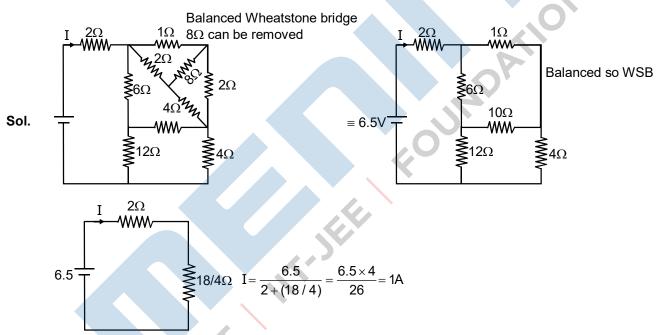
$$\frac{4}{1}\frac{1}{1}\frac{d\theta}{d\theta} = \frac{1}{1}\frac{8\sqrt{3}}{1} = \frac{4\sqrt{3}}{1}$$

$$\frac{d\theta}{dn} = 2$$
 m = 2 Ans.

2. In the following circuit, the current through the resistor R (=  $2\Omega$ ) is I Amperes. The value of I is



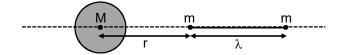
Ans. [1] current electricity



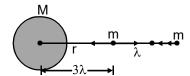
3. An electron in an excited state of  $Li^{2+}$  ion has angular momentum  $3h/2\pi$ . The de Broglie wavelength of the electron in this state is  $p\pi a_0$  (where  $a_0$  is the Bohar radius). The value of p is

Sol. 
$$L = \frac{nh}{2\pi} = \frac{3h}{2\pi} \Rightarrow n = 3$$
 
$$mvr = \frac{nh}{2\pi}$$
 
$$P = \frac{nh}{2\pi r}$$
 
$$So, \lambda = \frac{h}{P} = \frac{2\pi r}{n} = \frac{2\pi}{3} \left(\frac{n^2}{2}a_0\right) \setminus \frac{2\pi \times 3a_0}{3} = 2\pi a_0 = P\pi a_0$$
 
$$So. P = 2$$

4. A large spherical mass M is fixed at one position and two identical point masses m are kept on a line passing through the centre of M (see figure). The point masses are connected by a rigid massless rod of length  $\ell$  and this assembly is free to move along the line connecting them. All three masses interact only through their mutual gravitational interaction. When the point mass nearer to M is at a distance  $r = 3\ell$  from M, the tension in the rod is zero for m = k  $\left(\frac{M}{288}\right)$ . The value of k is:



Ans. [7]



Sol.

For tension in the rod to be zero.

Both masses should move with same acceleration, only due to gravitation alteraction

$$\frac{GMm}{9\ell^2} - \frac{Gm^2}{\ell^2} = \frac{GMm}{16\ell^2} + \frac{Gm^2}{\ell^2}$$

$$\frac{GMm}{\ell^2} \left\{ \frac{7}{9 \times 16} \right\} = \frac{2Gm^2}{\ell^2}$$

$$m = \left(\frac{7M}{288}\right) = K\frac{M}{288} \text{ SoK} = 7$$

5. The energy of a system as a function of time t is given as  $E(t) = A^2 \exp(-\alpha t)$ , where  $\alpha = 0.2 \text{ s}^{-1}$ . The measurement of A has an error of 1.25%. If the error in the measurement of time is 1.50%, the percentage error in the value of E(t) at t = 5 s is

Ans. [4

Sol. 
$$E(t) = A^{2}e^{-\alpha t}$$

$$dE = 2Ae^{-\alpha t} dA + A^{2} (-a)e^{-\alpha t} . dt$$

$$\frac{dE}{R} = \frac{2Ae^{-\alpha t} dA - \alpha A^{2}e^{-\alpha t} dt}{A^{2}e^{-\alpha t}}$$

$$\frac{2A}{A^2} = dA - \frac{\alpha A^2}{A^2} dt.$$

$$\frac{dE}{E} = \frac{2}{A} dA - \alpha dt$$

$$\frac{dE}{E} = \frac{2}{A}dA + \alpha dt = \text{(for errors)}$$

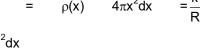
$$= 2 \times \frac{1.25}{100} + 0.2 \times 5 \times \frac{1.50}{100}$$

$$=\frac{2.5}{100}+\frac{1.50}{100}=\frac{4}{100}=4\%$$

The densities of two solid spheres A and B of the same radii R vary with radial distance r as  $\rho_A(r) = k \left(\frac{r}{R}\right)^{-1}$ 6. and  $\rho_B(r) = k \left(\frac{r}{R}\right)^{\circ}$ , respectively, where k is a constant. The moments of inertia of the individual spheres about axes passing through their centres are  $I_A$  and  $I_B$ , respectively. If  $\frac{I_B}{I_A} = \frac{n}{10}$ , the value of n is

Ans. [6]

$$4\pi x^2 dx$$



$$dI_A = \frac{2}{3} (dm)x^2 = \frac{2}{3} \frac{k}{R} 4\pi x^5 dx$$

$$dI_B = \frac{2}{3} \frac{K}{R^5} 4\pi x^9 dx$$

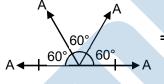
$$\frac{I_{_B}}{I_{_A}} = \frac{1}{R^4} \cdot \frac{R^{10}}{10} \cdot \frac{6}{R^6} = \frac{6}{10} = \frac{n}{10}$$

$$n = 6$$

Four harmonic waves of equal frequencies and equal intensities  $I_0$  have phase angles 0,  $\pi/3$ ,  $2\pi/3$  and 7.  $\pi$ . When they are superposed, the intensity of the resulting wave is  $nl_0$ . The value of n is

Ans. [3]

Sol.



$$I_0 \propto A^2$$

So, IR 
$$\propto (\sqrt{3} \text{ A})^2 = 3I_0$$

$$n = 3$$

For a radioactive material, its activity A and rate of change of its activity R are defined as  $A = -\frac{dN}{dt}$ 8. and R =  $-\frac{dA}{dt}$ , where N(t) is the number of nuclei at time t. Two radioactive sources P (mean life  $\tau$ ) and Q (mean life  $2\tau$ ) have the same activity at t = 0. Their rates of change of activities at t =  $2\tau$  are  $R_P$ and  $R_Q$ , respectively. If  $\frac{R_P}{R_Q} = \frac{n}{e}$ , then the value of n is

Ans. [2]

**Sol.** 
$$N = N_0 e^{-\lambda t}$$

$$A = -\frac{dN}{dt} = \lambda N_0 e^{-\lambda t} = \lambda N$$

$$R = -\frac{dA}{dt} = +\lambda^2 N_0 e^{-\lambda t} = \lambda A = \lambda^2 N = \lambda^2 N_0 e^{-\lambda t}$$

Now, 
$$\lambda_P N_{P0} = \lambda_O N_0$$

$$\frac{R_P}{R_Q} = \frac{\lambda_P}{\lambda_Q} \frac{A_P}{A_V} = 2$$

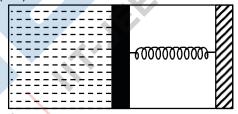
$$=2\frac{e^{-2}}{e^{-1}}=\frac{2}{e}$$

$$n = 2$$

#### **SECTION 2**

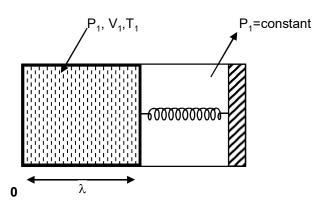
(Maximum Marks: 32)

- This section contains EIGHT questions
- Each questions has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is (are) correct
- For each question, darken the bubble(s) corresponding to all the correct option(s) in the ORS
- Marking scheme:
  - +4 If only the bubble(s) corresponding to all the correct option(s) is (are) darkened
  - 0 If none of the bubbles is darkened
  - -2 In all other cases
- 9. An ideal monoatomic gas is confined in a horizontal cylinder by a spring loaded piston (as shown in the figure). Initially the gas is at temperature T<sub>1</sub>, pressure P<sub>1</sub> and volume V<sub>1</sub> and the spring is in its relaxed state. The gas is then heated very slowly to temperature T<sub>2</sub>, pressure P<sub>2</sub> and volume V<sub>2</sub>. During this process the piston moves out by a distance x. Ignoring the friction between the piston and the cylinder, the correct statement(s) is (are)



- (A) If  $V_2 = 2V_1$  and  $T_2 = 3T_1$ , then the energy stored in the spring is  $\frac{1}{4}P_1V_1$
- (B) If  $V_2 = 2V_1$  and  $T_2 = 3T_1$ , then the change in internal energy is  $3P_1V_1$
- (C) If  $V_2 = 3V_1$  and  $T_2 = 4T_1$ , then the work done by the gas is  $\frac{7}{3}P_1V_1$
- (D) If  $V_2 = 3V_1$  and  $T_2 = 4T_1$ , then the heat supplied to the gas is  $\frac{17}{6} P_1 V_1$

Ans. [A, B, C]



Sol.

 $V_1$  = Al,  $\ell$  = initial length of segment containing monoatomic gas.

$$P_2 = \frac{T_2}{V_2} \frac{P_1 V_1}{T_1}$$

(A) 
$$P_2 = \frac{3T_1}{2V_1} \frac{P_1 V_1}{T_1} = \frac{3}{2} P_1$$
  
 $P_1 A + Kx = P_2 A$ 

$$kx = (P_2 - P_1)A = \left(\frac{3}{2} - 1\right)P_1A = \frac{P_1A}{2}$$

as 
$$v_2 = 2v_1$$
 so  $x = \ell$ 

so 
$$k\ell = \left(\frac{P_1A}{2}\right)$$

$$P_{1}A + Kx = P_{2}A$$

$$kx = (P_{2} - P_{1})A = \left(\frac{3}{2} - 1\right)P_{1}A = \frac{P_{1}A}{2}$$

$$as v_{2} = 2v_{1} \text{ so } x = \ell$$

$$so k\ell = \left(\frac{P_{1}A}{2}\right)$$

$$energy of spring = \frac{1}{2} kx^{2} = \frac{1}{2} k\ell^{2} = \frac{1}{2} \frac{P_{1}A\ell}{2} = \frac{P_{1}V_{1}}{4}$$

$$increase in U, \Delta U = \frac{3}{2} nR (2T_{1}) = 3nRT_{1} = 3P_{1}V_{1}$$

$$Hence q = \Delta U + w = 3P_{1}V_{1} + \frac{P_{1}V_{1}}{4} = \frac{13P_{1}V_{1}}{4}$$

increase in U, 
$$\Delta U = \frac{3}{2} \text{ nR } (2T_1) = 3 \text{nRT}_1 = 3 P_1 V_1$$

Hence 
$$q = \Delta U + w = 3P_1V_1 + \frac{P_1V_1}{4} = \frac{13P_1V_1}{4}$$

(C) For 
$$v_2 = 3v_1$$
 and  $T_2 = 4T_1$ 

$$P_2 = \frac{4}{3} P_1$$
  $\Rightarrow kx = \left(\frac{4}{3} - 1\right) P_1 A = \frac{P_1 A}{3}$ 

work done by gas = P1 (2
$$\ell$$
)A +  $\frac{1}{2}\frac{P_1A}{3}$  (2 $\ell$ ) = 2 $P_1v_1$  +  $\frac{P_1v_1}{3}$  =  $\frac{7}{3}$  =  $P_1V_1$ 

- A fission reaction is given by  $^{236}_{92}U \rightarrow^{140}_{54}Xe + ^{94}_{38}Sr + x + y$ , where x and y are two particles. Considering 10.  $_{92}^{236}\text{U}$  to be at rest, the kinetic energies of the products are denoted by  $K_{\text{Xe}}$ ,  $K_{\text{Sr}}$ ,  $K_{\text{x}}$  (2MeV) and  $K_{\text{y}}$  (2MeV), respectively. Let the binding energies per nucleon of  $^{236}_{92}$ U, $^{140}_{54}$  Xe , and  $^{94}_{38}$ Sr be 7.5 MeV, 8.5 MeV and 8.5 MeV, respectively. Considering different conservation laws, the correct option(s) is (are)
  - (A) x = n, y = n,  $K_{Sr} = 129$  MeV,  $K_{Xe} = 86$  MeV (B) x = p,  $y = e^-$ ,  $K_{Sr} = 129$  MeV,  $K_{Xe} = 86$  MeV
  - (C) x = p, y = n,  $K_{Sr} = 129$  MeV,  $K_{Xe} = 86$  MeV (D) x = n, y = n,  $K_{Sr} = 86$  MeV,  $K_{Xe} = 129$  MeV

Ans. [A]

**Sol.** 
$$_{92}U^{236} \longrightarrow _{54}Xe^{140} + _{38}Sr^{94} + x + y$$

(A) 
$$Q = (8.5 \times 140 + 8.5 \times 94 - 7.5 \times 236) \text{ MeV}$$
  
=  $(1989 - 1770) \text{ MeV} = 219 \text{ MeV}$ 

then 
$$K_{xe} + K_{Sr} + k_x + K_v = 219 \text{ MeV}$$

$$K_{x_e} + X_{s_r} = 219 - 4 = 215 \text{ MeV}$$

- (B) Not satisfying mass conservation and charge conservation.
- (C) Not satisfying charge conservation.
- (D) Will satisfy all conservation lenz except.

Momentum conservation for D

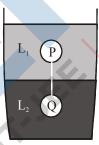
$$P_{xe} = \sqrt{2 \times 140 \times 129} = \sqrt{36120} = 190.05$$

$$P_{Sr} = \sqrt{2 \times 94 \times 86} = \sqrt{16168} = 127.15$$

$$P_n = \sqrt{2 \times 1 \times 2} = 2$$

Momentum conservation can not be satisfied in last case, so only (A)

11. Two spheres P and Q of equal radii have densities  $\rho_1$  and  $\rho_2$ , respectively, The spheres are connected by a massless string and placed in liquids  $L_1$  and  $L_2$  of densities  $\sigma_1$  and  $\sigma_2$  and viscosities  $\eta_1$  and  $\eta_2$ , respectively. They float in equilibrium with the sphere P in L<sub>1</sub> and sphere Q in L<sub>2</sub> and the string being taut (see figure). If sphere P alone in L<sub>2</sub> has terminal velocity  $\vec{V}_P$  and Q alone in L<sub>1</sub> has terminal velocity  $\vec{V}_Q$ , then



(A) 
$$\frac{|\overrightarrow{V}_P|}{|\overrightarrow{V}_Q|} = \frac{\eta_1}{\eta_2}$$

(B) 
$$\frac{|V_P|}{|\overrightarrow{V}_Q|} = \frac{\eta_2}{\eta_1}$$

(C) 
$$\overrightarrow{V}_P \cdot \overrightarrow{V}_Q > 0$$

(C) 
$$\overrightarrow{V}_P.\overrightarrow{V}_Q > 0$$
 (D)  $\overrightarrow{V}_P.\overrightarrow{V}_Q < 0$ 

**Sol.** 
$$T + v\rho_1 g = v\sigma_1 g$$

$$T = v(\sigma_1 - \rho_1)g$$

= 
$$v (\rho_2 - \sigma_2)g$$
 ......(ii

$$\sigma_1 - \rho_1 = \rho_2 - \sigma_2$$

$$\sigma_1 + \sigma_2 = \rho_1 + \rho_2$$

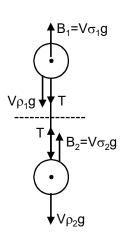
$$6\pi\eta_2 rv_P = mg - B = (v\rho_1 g - v\sigma_2 g)$$

= 
$$v(\rho_1 - \sigma_2)g$$

$$6\pi\eta_1\rho V_Q = mg - B = V(\rho_2 - \sigma_1)g$$

So, 
$$6\pi\eta_2 rv_P = -6\pi\eta_1 rvQ$$

So, 
$$\frac{|\mathbf{v}_p|}{|\mathbf{v}_Q|} = \frac{\mathbf{n}_1}{\mathbf{n}_2}$$
 and  $\vec{\mathbf{v}}_p \bullet \vec{\mathbf{v}}_Q < 0$ 



12. In terms of potential difference V, electric current I, permittivity  $\varepsilon_0$ , permeability  $\mu_0$  and speed of light c, the dimensionally correct equations is (are)

(A) 
$$\mu_0 I^2 = \varepsilon_0 V^2$$

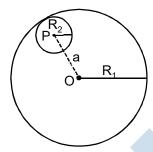
(B) 
$$\varepsilon_0 I = \mu_0 V$$

(C) 
$$I = \varepsilon_0 cV$$

(D) 
$$\mu_0 cI = \epsilon_0 V$$

Ans. [A, C]

13 Consider a uniform spherical charge distribution of radius R<sub>1</sub> centred at the origin O. In this distribution, a spherical cavity of radius  $R_2$ , centred at P with distance OP = a =  $R_1 - R_2$  (see figure) is made. If the electric field inside the cavity at position  $\vec{r}o = \vec{E}(\vec{r})$ , then the correct statement(s) is (are)



(A)  $\vec{E}$  is uniform, its magnitude is independent of  $R_2$  but its direction depends on T

(B)  $\vec{E}$  is uniform, its magnitude depends on  $R_2$  and its direction depends on T

(C) E is uniform, its magnitude is independent of a but its direction depends on a

(D)  $\vec{E}$  is uniform and both its magnitude and direction depend on  $\vec{a}$ 

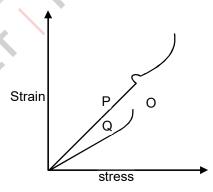
Ans.

**Sol.** 
$$\vec{E}_{cavity} = \frac{\rho}{3\epsilon_0} \vec{a}$$
  $a = (R_1 - R_2)$ 

$$a = (R_1 - R_2)$$

= uniform  $\vec{E}$ , magnitude and direction depends on a

14. In plotting stress versus strain curves for two materials P and Q, a student by mistake puts strain on the y-axis and stress on the x-axis as shown in the figure. Then the correct statement(s) is(are)



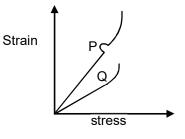
(A) P has more tensile strength than Q

(B) P is more ductile than Q

(C) P is more brittle than Q

(D) The Young's modulus of P is more than that of Q

Ans. [A, B] Sol.



Strain O P stress

(A)

Q has more tensile strength

P is more ductile than Q(B)

Y of Q is more than P, So (D) is incorrect

15. A spherical body of radius R consists of a fluid of constant density and is in equilibrium under its own gravity. If P(r) is the pressure at r(r < R), then the correct option(s) is (are)

(A) 
$$P(r = 0) = 0$$

(B) 
$$\frac{P(r = 3R/4)}{P(r = 2R/3)} = \frac{63}{80}$$

(C) 
$$\frac{P(r = 3R/5)}{P(r = 2R/5)} = \frac{16}{21}$$

(D) 
$$\frac{P(r = R/2)}{P(r = R/3)} = \frac{20}{27}$$

Ans. [B, C]

**Sol.** 
$$g = \frac{GM}{R^3} r$$

$$dF = -(dm)g = \frac{GM}{R^3} \rho dA r dr$$

$$dP = -\frac{dF}{dA} = -\frac{GM}{R^3}\rho r dr$$

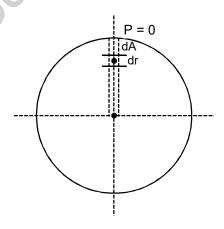
$$\int_{P}^{0} dP = -\frac{GM}{R^{3}} \int_{r}^{R} \rho r dr$$

$$+P = +\frac{GM}{R^3}\frac{\rho}{2} (R^2 - r^2)$$

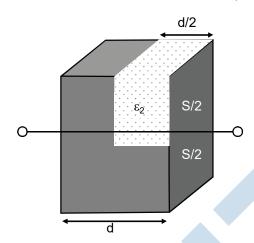
$$\frac{P(r = 3R/4)}{P(r = 2R/3)} = \frac{\left(R^2 - (9/16)R^2\right)}{\left(R^2 - (4/9)R^2\right)} = \frac{7}{16} \times \frac{9}{5} = \frac{63}{80}$$

(C) 
$$\frac{P(r=3R/5)}{P(r=2R/5)} = \frac{(R^2 - (9/25)R^2)}{(R^2 - (4/25)R^2)} = \frac{16}{25} \times \frac{25}{21} = \frac{16}{21}$$

(D) 
$$\frac{P(r=R/2)}{P(r=R/2)} = \frac{(R^2 - R^2/4)}{(R^2 - R^2/9)} = \frac{3/4}{8/9} = \frac{3}{4} \times \frac{9}{8} = \frac{27}{32}$$



A parallel plate capacitor having plates of area S and plate separation d, has capacitance  $C_1$  in air. When two dielectrics of different relative permittivities ( $\epsilon_1$  = 2 and  $\epsilon_2$  = 4) are introduced between the two plates as shown in the figure, the capacitance becomes  $C_2$ . The ratio  $\frac{C_2}{C_1}$  is



(A) 6/5

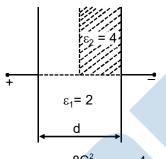
(B) 5/3

(C) 7/5

(D) 7/3

Ans. [D]

**Sol.**  $C_1 = \frac{\varepsilon_0 S}{d}$ 



 $C_{eq} = C_1 + \frac{8C_1^2}{6C_1} = C_1 + \frac{4}{3} C_1 = \frac{7}{3} C_1$ 

# **SECTION 3**

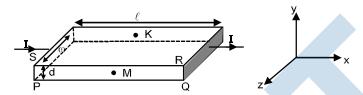
(Maximum Marks: 16)

- · This section contains TWO paragraphs
- · Based on each paragraph, there will be TWO questions
- Each question has FOUR options (A), (B), (C) and (D). One or more than one of these four option(s) is(are) correct
- For each question darken the bubble(s) corresponding to all the correct option(s) is the ORS
- Marking scheme:
  - + 4 If only the bubble (s) corresponding to all the correct option(s) is/are darkened
  - 0 If none of the bubbles is darkened
  - 2 In all other cases

#### **PARAGRAPH 1**

In a thin rectangular metallic strip a constant current I flows along the positive x-direction, as shown in the figure. The length, width and thickness of the strip are I, w and d, respectively.

A uniform magnetic field  $\tilde{B}$  is applied on the strip along the positive y-direction. Due to this, the charge carriers experience a net deflection along the z-direction. This results in accumulation of charge carriers on the surface PQRS and appearance of equal and opposite charges on the face opposite to PQRS. A potential difference along the z-direction is thus developed. Charge accumulation continues until the magnetic force is balanced by the electric force. The current is assumed to be uniformly distributed on the cross section of the strip and carried by electrons.



- 17. Consider two different metallic strips (1 and 2) of the same material. Their lengths, are the same, width are w<sub>1</sub> and w<sub>2</sub> and thickness are d<sub>1</sub> and d<sub>2</sub> respectively. Two points K and M are symmetrically located on the opposite faces parallel to the x-y plane (see figure). V<sub>1</sub> and V<sub>2</sub> are the potential differences between K and M in strips 1 and 2, respectively. Then, for a given current I flowing through them in a given magnetic field strength B, the correct statement(s) is (are).
  - (A) If  $w_1 = w_2$  and  $d_1 = 2d_2$ , then  $V_2 = 2V_1$
  - (B) If  $w_1 = w_2$  and  $d_1 = 2d_2$ , then  $V_2 = V_1$
  - (C) If  $w_1 = 2w_2$  and  $d_1 = d_2$ , then  $V_2 = 2V_1$
  - (D) If  $w_1 = 2w_2$  and  $d_1 = d_2$ , then  $V_2 = V_1$
- Ans. [A, D]
- Sol.  $\frac{I}{ne\omega d}B = \frac{V}{w}$   $v = \left(\frac{IB}{ned}\right)$   $\frac{v_1}{v_2} = \frac{d_2}{d_1} = \frac{1}{2} \text{ So, } v_2 = 2v_1$
- 18. Consider two different metallic strips (1 and 2) of same dimensions (length *I*, width ω and thickness d) with carrier densities n<sub>1</sub> and n<sub>2</sub>, respectively. Strip 1 is placed in magnetic field B<sub>1</sub> and strip 2 is placed in magnetic field B<sub>2</sub>, both along positive y-directions. Then V<sub>1</sub> and V<sub>2</sub> are the potential differences developed between K and M is strips 1 and 2, respectively. Assuming that the current I is the same for both the strips, the correct option(s) is (are)
  - (A) If  $B_1 = B_2$  and  $n_1 = 2n_2$ , then  $V_2 = 2V_1$
  - (B) If  $B_1 = B_2$  and  $n_1 = 2n_2$ , then  $V_2 = V_1$
  - (C) If  $B_1 = 2B_2$  and  $n_1 = n_2$ , then  $V_2 = 0.5V_1$
  - (D) If  $B_1 = 2B_2$  and  $n_1 = n_2$ , then  $V_2 = V_1$
- Ans. [A, C]

Sol. 
$$ev_{d_1}B_1 = e\frac{v_1}{\omega}$$

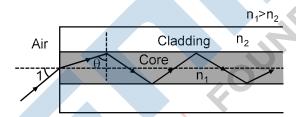
$$\Rightarrow v_1 = \frac{\omega IB_1}{n_1 ewd} = \frac{IB_1}{n_1 de}$$

$$\frac{v_1}{v_2} = \frac{B_1}{n_1} \frac{n_2}{B_2} = \frac{1}{2} = = \Rightarrow v_2 = 2v_1$$

$$\frac{v_1}{v_2} = 2 \Rightarrow v_2 = 0.5 v_1$$

#### **PARAGRAPH 2**

Light guidance in an optical fiber can be understood by considering a structure comprising of thin solid glass cylinder of refractive index  $n_1$  surrounded by a medium of lower refractive index  $n_2$ . The light guidance in the structure takes place due to successive total internal reflections at the interface of the media  $n_1$  and  $n_2$  as shown in the figure. All rays with the angle of incidence i less than a particular value  $i_m$  are confined in the medium of refractive index  $n_1$ . The numerical aperture (NA) of the structure is defined as  $\sin i_m$ .



- 19. For two structures namely  $S_1$  with  $n_1 = \sqrt{45} / 4$  and  $n_2 = 3/2$ , and  $S_2$  with  $n_1 = 8/5$  and  $n_2 = 7/5$  and taking the refractive index of water to be 4/3 and that of air to be 1, the correct option(s) is(are)
  - (A) NA of S<sub>1</sub> immersed in water is the same as that of S<sub>2</sub> immersed in a liquid of refractive index  $\frac{16}{3\sqrt{15}}$
  - (B) NA of S<sub>1</sub> immersed in liquid of refractive index  $\frac{16}{\sqrt{15}}$  is the same as that of S<sub>2</sub> immersed in water.
  - (C) NA of S<sub>1</sub> placed in air is the same as that of S<sub>2</sub> immersed in liquid of refractive index  $\frac{4}{\sqrt{15}}$ .
  - (D) NA of S<sub>1</sub> placed in air is the same as that of S<sub>2</sub> placed in water.

Ans. [A, C]

$$\begin{aligned} \text{Sol.} \qquad & \sin C \ = \frac{\mu_r}{\mu_d} = \frac{n_2}{n_1} \\ & \mu_s \sin i = n \sin r = n_1 \sin \left( 90 - Q \right) \\ & \mu_s \sin i = n_1 \cos Q = n1 \sqrt{1 - \frac{n_2^2}{n_1^2}} = \sqrt{n_1^2 - n_2^2} \\ & \sin i = \frac{1}{H} \sqrt{n_1^2 - n_2^2} \end{aligned}$$

(A) 
$$S_{1}: \sin i_{m_{1}} = \frac{3}{4} \sqrt{\frac{45}{16} - \frac{9}{4}} = \frac{3}{4} \sqrt{\frac{45 - 36}{16}} = \frac{3}{4} \times \frac{3}{4} = \frac{9}{16}$$

$$S_{2}: \sin i_{m_{2}} = \frac{3\sqrt{15}}{16} \sqrt{\frac{64}{25} - \frac{49}{25}} = \frac{3\sqrt{15}}{16} \frac{\sqrt{15}}{5} = \frac{9}{16}$$

(B) 
$$S_{1}: \sin i_{m_{1}} = \frac{\sqrt{15}}{6} \sqrt{\frac{45}{16} - \frac{9}{4}} = \frac{\sqrt{15}}{6} \sqrt{\frac{45 - 36}{16}} = \frac{3}{4} \frac{\sqrt{15}}{6} = \frac{\sqrt{15}}{8}$$

$$S_{2}: \sin i_{m_{2}} = \frac{3}{4} \sqrt{\frac{64}{25} - \frac{49}{25}} = \frac{3}{4} \sqrt{\frac{15}{25}} = \frac{\sqrt{15} \times 3}{20}$$

(C) 
$$S_{1} : \sin i_{m_{1}} = \sqrt{\frac{45}{16} - \frac{9}{4}} = \frac{\sqrt{9}}{4} = \frac{3}{4}$$

$$S_{2} : \sin i_{m_{2}} = \frac{\sqrt{15}}{4} \sqrt{\frac{15}{25}} = \frac{15}{4 \times 5} = \frac{15}{20} = \frac{3}{4}$$

(D) 
$$S_1 : \sin i_{m_1} = \frac{3}{4}$$
  $S_2 : \sin i_{m_2} = \frac{\sqrt{15}}{5}$ 

20. If two structures of same cross-sectional area, but different numerical apertures  $NA_1$  and  $NA_2$  ( $NA_2 < NA_1$ ) are joined longitudinally, the numerical aperture of the combined structure is

$$(A) \ \frac{NA_1NA_2}{NA_1 + NA_2}$$

- (B) NA<sub>1</sub> + NA<sub>2</sub>
- (C) NA
- (D) NA<sub>2</sub>

Ans. [D]

## **PART B: CHEMISTRY**

(Maximum Marks: 32)

- This section contains Eight questions.
- The answer to each question is a SINGLE DIGIT INTEGER ranging from 0 to 9, both inclusive.
- For each question, darken the bubble corresponding to the correct integer in the ORS.
- Marking Scheme
  - + 4 If the bubble corresponding to the answer is darkened.
  - 0 In all other cases.
- 21. The molar conductivity of a solution of a weak acid HX (0.01 M) is 10 times smaller than the molar conductivity of a solution of a weak acid HY (0.10 M). If  $\lambda_{X^-}^0 \approx \lambda_{Y^-}^0$ , the difference in their pK<sub>a</sub> values, pK<sub>a</sub>(HX) pK<sub>a</sub>(HY), is (consider degree of ionisation of both acids to be << 1)

Ans. [3]

**Sol.** HX □ H<sup>+</sup> + X<sup>-</sup>

$$Ka = \frac{\left[H^{+}\right]\left[X^{-}\right]}{\left[HX\right]}$$

$$Ka = \frac{\left[H^{+}\right]\left[Y^{-}\right]}{\left[HY\right]}$$

$$\Lambda_{\rm m}$$
 for HX =  $\Lambda_{\rm m}$ 

$$\Lambda_{\mathsf{m}}\mathsf{for}\,\mathsf{HY}=\Lambda_{\mathsf{m}_2}$$

$$\Lambda_{m_1} = \frac{1}{10} \Lambda_{m_2}$$

$$Ka = C\alpha^2$$

$$Ka_1 = C_1 \times \left(\frac{\Lambda_{m_1}}{\Lambda_{m_1}^0}\right)^2$$

$$\mathsf{Ka}_2 = \mathsf{C}_2 \times \left(\frac{\Lambda_{\mathsf{m}_2}}{\Lambda_{\mathsf{m}_0}^0}\right)^2$$

$$\frac{Ka_{_1}}{Ka_{_2}} = \frac{C_{_1}}{C_{_2}} \times \left(\frac{\Lambda_{_{m_{_1}}}}{\Lambda_{_{m_{_2}}}^0}\right)^2 = \frac{0.01}{0.1} \times \left(\frac{1}{10}\right)^2 = 0.001$$

 $pKa_1-pKa_2 = 3$ 

22. A closed vessel with rigid walls contains 1 mol of  $^{238}_{92}$ U and 1 mol of air at 298 K. Considering complete decay of  $^{238}_{92}$ U to  $^{206}_{92}$ Pb to , the ratio of the final pressure to the initial pressure of the system at 298K is

Ans. [9]

**Sol.** In conversion of  ${}^{238}_{92}$ U to  ${}^{206}_{92}$ Pb , 8  $\alpha$  - particles and 6  $\beta$  particles are ejected.

The number of gaseous moles initially = 1 mol

The number of gaseous moles finally = 1 + 8 mol; (1 mol from air and 8 mol of 2He<sup>4</sup>)

So the ratio = 9/1 = 9

- 23. In dilute aqueous  $H_2SO_4$ , the complex diaquodioxalatoferrate(II) is oxidized by  $MnO_4^-$ . For this reaction, the ratio of the rate of change of  $[H^+]$  to the rate of change of  $[MnO_4^-]$  is
- Ans. [8]
- **Sol.**  $\left[ Fe(C_2O_4)(H_2O) \right]^{2-} + MnO_4^{2-} + 8H^+ \longrightarrow Mn^{2+} + Fe^{3+} + Fe^{3+} + 4CO_2 + 6H_2O_3 +$

So the ratio of rate of change of [H $^+$ ]to that of rate  $\lceil MnO_4^- \rceil$  of is 8.

24. The number of hydroxyl group(s) in **Q** is

$$H \xrightarrow{H^+} P \xrightarrow{\text{aqueous dilute KMnO}_4(\text{excess})} Q$$
 $H \xrightarrow{\text{heat}} P \xrightarrow{\text{heat}} P$ 

Ans. [4]

aqueous dilute KMnO<sub>4</sub> (excess)

25. Among the following, the number of reaction(s) that produce(s) benzaldehyde is:

III. 
$$H_2$$
 Pd-BaSO<sub>4</sub>

$$CO_2Me$$
 DIBAL-H Toluene –  $78^{\circ}C$ 

Ans. [4]

II 
$$H_2O$$
 CHO

III 
$$H_2O$$
 CHO Pd-BaSO<sub>4</sub>

- 26. In the complex acetylbromidodicarbonylbis(triethylphosphine)iron(II), the number of Fe–C bond(s) is
- Ans. [3]

The number of – C bonds is 3.

- 27. Among the complex ions,  $[Co(NH_2 CH_2 CH_2 NH_2)_2Cl_2]^{\dagger}$ ,  $[CrCl_2(C_2O_4)_2]^{3-}$ ,  $[Fe(H_2O)_4(OH)_2]^{\dagger}$ ,  $[Fe(NH_3)_2(CN)_4]^{-}$ ,  $[Co(NH_2-CH_2 CH_2 NH_2)_2(NH_3)Cl_2^{2+}$  and  $[Co(NH_3)_4(H_2O)Cl_2^{2+}]$ , the number of complex ion(s) that show(s) cis-trans isomerism is:
- Ans. [6]
- **Sol.**  $[CO(en)_2Cl_2]^+\longrightarrow will$  show cis trans isomerism  $[CrCl_2(C_2O_4)_2]^{3-}\longrightarrow will$  show cis trans isomerism  $[Fe(H_2O)_4(OH)_2]^-\longrightarrow will$  show cis trans isomerism  $[Fe(CN)_4(NH_3)_2]^-\longrightarrow will$  show cis trans isomerism  $[Co(en)_2(NH_3)Cl]^{2+}\longrightarrow will$  show cis trans isomerism  $[Co(NH_3)_4(H_2O)Cl^{2+}]\longrightarrow will$  not show cis trans isomerism (Although it will show geometrical isomerism)
- **28.** Three moles of  $B_2H_6$  are completely reacted with methanol. The number of moles of boron containing product formed is:
- Ans. [6]
- **Sol.**  $B_2H_6 + 6MeOH \longrightarrow 2B (OMe)_3 + 6H_2$

1 mole of B<sub>2</sub>H<sub>6</sub> reacts with 6 mole of MeOH to give 2 moles of B(OMe)<sub>3</sub>.

3 mole of B<sub>2</sub>H<sub>6</sub> will react with 18 mole of MeOH to give 6 moles of B(OMe)<sub>3</sub>

## **SECTION 2**

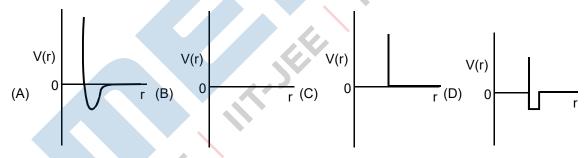
(Maximum Marks: 32)

- This section contains EIGHT questions
- Each question has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is(are) correct
- For each question, darken the bubble(s) corresponding to all the correct option(s) in the ORS
- Marking scheme:
  - +4 If only the bubble(s) corresponding to all the correct option(s) is (are) darkened
  - 0 If none of the bubbles is darkened
  - –2 In all other cases
- When  $O_2$  is adsorbed on a metallic surface, electron transfer occurs from the metal to  $O_2$ . The **TRUE** statement(s) regarding this adsorption is(are)
  - (A) O<sub>2</sub> is physisorbed

- (B) heat is released
- (C) occupancy of  $\dot{\sigma}_{2p}$  of  $O_2$  is increased
- (D) bond length of O<sub>2</sub> is increased.

#### Ans. [B], [C], [D]

- **Sol.** \* Adsorption of O<sub>2</sub> on metal surface is exothermic.
  - \* During electron transfer from metal to  $O_2$  electron occupies  $\pi^*_{20}$  orbital of  $O_2$ .
  - \* Due to electron transfer to O<sub>2</sub> the bond order of O<sub>2</sub> decreases hence bond length increases.
- 30. One mole of a monoatomic real gas satisfies the equation p(V b) = RT where b is a constant. The relationship of interatomic potential V(r) and interatomic distance r for the gas is given by:



- Ans. [C]
- **Sol.** At large inter-ionic distances (because  $a \rightarrow 0$ ) the P.E. would remain constant. However, when  $r \rightarrow 0$ ; repulsion would suddenly increase
- **31.** In the following reactions, the product **S** is

$$(A) \begin{array}{c} H_3C \\ \hline \\ H_3C \\ \hline \\ \end{array} \begin{array}{c} \text{i. O}_3 \\ \text{ii. Zn, H}_2O \end{array} R \begin{array}{c} NH_3 \\ \hline \\ \end{array} \begin{array}{c} S \\ \hline \\ \end{array}$$

$$(C)_{H_3C}$$
  $(D)_{H_3C}$ 

Ans. [A]

Ans.

Sol.

Sol. 
$$H_3C$$
  $(i) O_3$   $(i) Z_{n_1} H_2O$   $(i) O_3$   $(i) Z_{n_2} H_2O$   $(i) O_3$   $(i) Z_{n_1} H_2O$   $(i) O_3$   $(i) O_4$   $(i) O$ 

**32.** The major product **U** in the following reactions is

$$\begin{array}{c} CH_2 = CH - CH_3, H^{+} \\ \text{high pressure, heat} \end{array} \qquad T \xrightarrow{\text{radical intiator, O}_2} U$$

$$(A) \qquad \qquad (B) \qquad (CH_3) \qquad (CH_3) \qquad (CH_3) \qquad (CH_3) \qquad (CH_2) \qquad (CH_3) \qquad (CH_2) \qquad (CH_3) \qquad (CH$$

СН₃

H<sub>3</sub>C

33 In the following reactions, the major product **W** is

Ans. [A]

Sol. 
$$NH_{2} \longrightarrow NH_{2} \longrightarrow NH_{2$$

- **34.** The correct statement(s) regarding, (i) HClO, (ii) HClO<sub>2</sub>, (iii) HClO<sub>3</sub> and (iv) HClO<sub>4</sub>, is(are)
  - (A) The number of CI = O bonds in (ii) and (iii) together is two
  - (B) The number of lone pairs of electrons on Cl in (ii) and (iii) together is three
  - (C) The hybridization of Cl in (iv) is sp<sup>3</sup>
  - (D) Amongst (i) to (iv), the strongest acid is (i)

Ans. [B], [C]

- 35. The pair(s) of ions where BOTH the ions are precipitated upon passing H<sub>2</sub>S gas in presence of dilute HCI, is(are)
  - (A) Ba<sup>2+</sup>, Zn<sup>2+</sup>
- (B) Bi<sup>3+</sup>, Fe<sup>3+</sup>
- (C) Cu<sup>2+</sup>, Pb<sup>2+</sup>
  - (D) Hg<sup>2+</sup>, Bi<sup>3+</sup>

Ans. [C], [D]

**Sol.**  $Cu^{2+}$ ,  $Pb^{2+}$ ,  $Hg^{2+}$ ,  $Bi^{3+}$  give ppt with  $H_2S$  in presence of dilute HCI.

- **36.** Under hydrolytic conditions, the compounds used for preparation of linear polymer and for chain termination, respectively, are
  - (A) CH<sub>3</sub>SiCl<sub>3</sub> and Si(CH<sub>3</sub>)<sub>4</sub>

(B) (CH<sub>3</sub>)<sub>2</sub>SiCl<sub>2</sub> and (CH<sub>3</sub>)<sub>3</sub>SiCl

© (CH<sub>3</sub>)<sub>2</sub>SiCl<sub>2</sub> and CH<sub>3</sub>SiCl<sub>3</sub>

(D) SiCl<sub>4</sub> and (CH<sub>3</sub>)<sub>3</sub>SiCl

Ans. [B]

#### **SECTION 3**

(Maximum Marks: 16)

- This section contains TWO paragraphs
- . Based on each paragraph, there will be TWO questions
- . Each question has FOUR options (A), (B), (C) and (D). One or more than one of these four option(s) is(are) correct
- . For each question darken the bubble(s) corresponding to all the correct option(s) in the ORS
- . Marking scheme:
  - + 4 If only the bubble (s) corresponding to all the correct option(s) is/are darkened
  - 0 If none of the bubbles is darkened
  - 2 In all other cases

#### PARAGRAPH - 1

In the following reactions:

$$\begin{array}{c} C_8H_6 \xrightarrow{Pd-BaSO_4} C_8H_8 \xrightarrow{i.B_2H_6} XX \\ \downarrow H_2O \\ \downarrow HgSO_4, H_2SO_4 \\ C_8H_8O \xrightarrow{i.EtMgBr, H_2O} Y \end{array}$$

37. Compound X is

$$(A)$$
  $(B)$   $(C)$   $(D)$   $(D)$   $(D)$ 

Ans. [C]

**38.** The major compound Y is

Ans. [D]

## PARAGRAPH - 2

When 100 mL of 1.0 M HCl was mixed with 100 mL of 1.0 M NaOH in an insulated beaker at constant pressure, a temperature increase of 5.7°C was measured for the beaker and its contents (**Expt. 1**). Because the enthalpy of neutralization of a strong acid with a strong base is a constant (– 57.0 kJ mol<sup>-1</sup>), this experiment could be used to measure the calorimeter constant.

In a second experiment (**Expt. 2**), 100 mL of 2.0 M acetic acid ( $K_a = 2.0 \times 10^{-5}$ ) was mixed with 100 mL of 1.0 M NaOH (under identical conditions of **Expt. 1**) where a temperature rise of 5.6°C was measured.

(Consider heat capacity of all solutions as 4.2 J  $\mathrm{g}^{-1}\mathrm{K}^{-1}$  and density of all solutions as 1.0 g  $\mathrm{mL}^{-1}$ )

- 39. Enthalpy of dissociation (in kJ mol<sup>-1</sup>) of acetic acid obtained from the Expt. 2 is
  - (A) 1.0
- (B) 10.0
- (C) 24.5
- (D) 51.4

Ans. [A]

**Sol.** HCl + NaOH  $\longrightarrow$  NaCl + H<sub>2</sub>O

 $n = 100 \times 1 = 100 \text{ m mole} = 0.1 \text{ mole}$ 

Energy evolved due to neutralization of HCl and NaOH = 0.1 × 57 = 5.7 kJ = 5700 Joule

Energy used to increase temperature of calorimeter = 5700 – 4788 = 912 Joule

ms. 
$$\Delta t = 912$$

$$m.s \times 5.7 = 912$$

ms = 160 Joule/°C [Calorimeter constant]

Energy evolved by neutralization of CH<sub>3</sub>COOH and NaOH

$$= 200 \times 4.2 \times 5.6 + 160 \times 5.6 = 5600$$
 Joule

So energy used in dissociation of 0.1 mole CH3COOH = 5700  $\square$  5600 = 100 Joule

Enthalpy of dissociation = 1 kJ/mole

40. The pH of the solution after Expt.2 is

Ans. [B]

**Sol.** 
$$CH_3COOH = \frac{1 \times 100}{200} = \frac{1}{2}$$

$$CH_3CONa = \frac{1 \times 100}{200} = \frac{1}{2}$$

$$pH = pK_a + log \frac{\left[salt\right]}{\left[acid\right]}$$

$$pH = 5 - log 2 + log \frac{1/2}{1/2}$$

# PART C: MATHEMATICS

Section 1 (Maximum Marks: 32)

- This section contains Eight questions.
- The answer to each question is a SINGLE DIGIT INTEGER ranging from 0 to 9, both inclusive.
- For each question, darken the bubble corresponding to the correct integer in the ORS.
- Marking Scheme
  - If the bubble corresponding to the answer is darkened.
  - In all other cases.
- If  $\alpha = \int_{2}^{1} (e^{9x+3\tan^{-1}x}) \left(\frac{12+9x^2}{1+x^2}\right) dx$  where  $\tan^{-1}x$  takes only principal values, then the value of 41.

$$\left(\log_e \left| 1 + \alpha \right| - \frac{3\pi}{4}\right)$$
 is

- Ans.
- $\alpha = \int_{0}^{1} (e^{9x + 3\tan^{-1}x}) \left( \frac{12 + 9x^{2}}{1 + x^{2}} \right) dx$ Sol.

Let 
$$9x + 3tan^{-1}x = t$$

$$\Rightarrow \left(9 + \frac{3}{1 + x^2}\right) dx = dt$$

$$\begin{split} \alpha &= \int\limits_0^{\left(9+\frac{3\pi}{4}\right)} e^t \ dt = \left(e^t\right)_0^{\left(9+\frac{3\pi}{4}\right)} = e^{\left(9+\frac{3\pi}{4}\right)} - 1 \\ \therefore 1 + \alpha &= e^{\left(9+\frac{3\pi}{4}\right)} \\ \Rightarrow \log_e \mid 1 + \alpha \mid = 9 + \frac{3\pi}{4} \\ \Rightarrow \log_e \mid 1 + \alpha \mid -\frac{3\pi}{4} = 9. \ \text{Ans.} \end{split}$$

$$\therefore 1 + \alpha = e^{\left(9 + \frac{3\pi}{4}\right)}$$

$$\Rightarrow \log_{\rm e} |1 + \alpha| = 9 + \frac{3\pi}{4}$$

$$\Rightarrow \log_e |1 + \alpha| - \frac{3\pi}{4} = 9$$
. Ans.

Let f: R o R be a continuous odd function, which vanishes exactly at one point and f(1) =  $\frac{1}{2}$ . 42.

Suppose that  $F(x) = \int_{1}^{x} f(t) dt$  for all  $x \in [-1, 2]$  and  $G(x) = \int_{1}^{x} t |f(f(t))| dt$  for all  $x \in [-1, 2]$ .

If = 
$$\lim_{x\to 1} \frac{F(x)}{G(x)} = \frac{1}{14}$$
, then the value of  $f\left(\frac{1}{2}\right)$  is

**Sol.** 
$$f(-x) = -f(x)$$

:. 
$$f(0) = 0$$
 and  $f(1) = \frac{1}{2}$ 

$$F(x) = \int_{-1}^{x} f(t) dt = F'(x) = f(x)$$

$$G(x) = \int_{-1}^{x} t \left| f(f(t)) \right| dt \Rightarrow G'(x) = x | f(f(x)) |$$

$$\lim_{x \to 1} \frac{F(x)}{G(x)} = \lim_{x \to 1} \frac{F'(x)}{G'(x)} = \frac{F'(1)}{G'(1)} = \frac{f(1)}{\left| f(f(1)) \right|} = \frac{\frac{1}{2}}{\left| f(\frac{1}{2}) \right|} = \frac{1}{14}$$

$$\Rightarrow \left| f\left(\frac{1}{2}\right) \right| = 7$$

f(x) vanishes at only one point

$$\therefore$$
  $f\left(\frac{1}{2}\right) = 7$ . Ans.

Suppose that  $\vec{p}$ ,  $\vec{q}$  and  $\vec{r}$  are three non-coplanar vectors in  $R^3$ . Let the components of a vector  $\vec{s}$  along  $\vec{p}$ ,  $\vec{q}$  and  $\vec{r}$  be 4, 3 and 5 respectively. If the components of this vector  $\vec{s}$  along

 $(-\vec{p}+\vec{q}+\vec{r}), (\vec{p}-\vec{q}+\vec{r})$  and  $(-\vec{p},-\vec{q}+\vec{r})$  are x, y and z respectively, then the value of 2x + y + z

$$\frac{\sum\limits_{k=1}^{12}\!\mid\alpha_{k+1}-\alpha_{k}\mid}{\sum\limits_{k=1}^{3}\!\mid\alpha_{4k-1}-\alpha_{4k-2}\mid}\text{ is }$$

Ans. Bonus

**44.** For any integer k, let  $\alpha_k = \cos\left(\frac{k\pi}{7}\right) + i\sin\left(\frac{k\pi}{7}\right)$ , where  $i = i = \sqrt{-1}$ . The value of the expression is

Ans. 4

**Sol.**  $\alpha_k = e^{\frac{ik\pi}{7}} = \alpha^k$  where  $\alpha = e^{\frac{i\pi}{7}}$  and  $|\alpha| = 1$ 

$$\frac{\sum\limits_{k=1}^{12}\mid\alpha_{k+1}-\alpha_{k}\mid}{\sum\limits_{k=1}^{3}\mid\alpha_{4k-1}-\alpha_{4k-2}\mid}=\frac{\sum\limits_{k=1}^{12}\mid\alpha\mid^{k}\mid\alpha-1\mid}{\sum\limits_{k=1}^{3}\mid\alpha\mid^{4k-2}\mid\alpha-1\mid}=\frac{\sum\limits_{k=1}^{12}\mid\alpha-1\mid}{\sum\limits_{k=1}^{3}\mid\alpha-1\mid}=\frac{12\mid\alpha-1\mid}{3\mid\alpha-1\mid}=4. \text{ Ans.}$$

45. Suppose that all the terms of an arithmetic progression (A.P.) are natural numbers. If the ratio of the sum of the first seven terms to the sum of the first eleven terms is 6: 11 and the seventh term lies in between 130 and 140, then the common difference of this A.P. is

**Sol.** 
$$\therefore \frac{S_7}{S_{11}} = \frac{\frac{7}{2}(2a+6d)}{\frac{11}{2}(2a+10d)} = \frac{6}{11}$$

$$\Rightarrow$$
 7 (2a + 6d) = 6 (2a + 10d)

$$\Rightarrow$$
 2a = 18d  $\Rightarrow$  a = 9d

$$T_7 = a + 6d$$

 $130 < a + 6d < 140 \Rightarrow 130 < 15d < 140$ 

$$\Rightarrow \frac{130}{15} < d < \frac{140}{15} \Rightarrow \frac{26}{3} < d < \frac{28}{3}$$

∴ d = 9. **Ans.** 

**46.** The Coefficient of  $x^9$  in the expansion of  $(1 + x)(1 + x^2)(1 + x^3).....(1 + x^{100})$  is

Ans. 8

**Sol.**  $x^9$  can be obtained by multiplying terms containing powers of x.

$$(0, 9), (1, 8), (2, 7), (3, 6), (4, 5), (1, 2, 6), (1, 3, 5), (2, 3, 4)$$

∴ coefficient of x<sup>9</sup> is 8. **Ans.**]

47. Suppose that the foci of the ellipse  $\frac{x^2}{9} + \frac{y^2}{5} = 1$  are  $(f_1, 0)$  and  $(f_2, 0)$  where  $f_1 > 0$  and  $f_2 < 0$ . Let  $P_1$  and

 $P_2$  be two parabolas with a common vertex at (0, 0) and with foci at  $(f_1, 0)$  and  $(2f_2, 0)$ , respectively. Let

 $T_1$  be a tangent to  $P_1$  which passes through (2 $f_2$ , 0) and  $T_2$  be a tangent to  $P_2$  which passes through ( $f_1$ ,

0). If  $m_1$  is the slope of  $T_1$  and  $m_2$  is the slope of  $T_2$ , then the value of  $\left(\frac{1}{m_1^2} + m_2^2\right)$  is

Ans. 4

**Sol.** e =  $\sqrt{1-\frac{5}{9}} = \frac{2}{3}$ 

∴ foci are (±2, 0)

 $f_1$ : (2, 0),  $f_2$ : (-2, 0)

Tangent to p<sub>1</sub>

 $y = m_1 x + \frac{2}{m_1}$ , which passes through (-4, 0)

$$0 = -4m_1 + \frac{2}{m_1} \implies m_1^2 = \frac{1}{2}$$

Tangent to p<sub>2</sub>

 $y = m_2 x - \frac{4}{m_2}$ , which passes through (2, 0)

$$0 = 2m_2 - \frac{4}{m_2} \Rightarrow m_2^2 = 2$$

$$\therefore \frac{1}{m_1^2} + m_2^2 = 2 + 2 = 4 \text{ Ans.}$$

**48.** Let m and n be two positive integers greater than 1. If  $\lim_{\alpha \to 0} \left( \frac{e^{\cos(\alpha^n)} - e}{\alpha^m} \right) = -\left( \frac{e}{2} \right)$  then the value of  $\frac{m}{n}$ 

is

**Sol.** 
$$\lim_{\alpha \to 0} \frac{e^{\cos(\alpha^n)} - e}{\alpha^m} = -\frac{e}{2}$$

$$\lim_{\alpha \to 0} \frac{e^{\cos(\alpha^n)-1}-1}{\alpha^m} = \frac{-1}{2}$$

$$\lim_{\alpha \to 0} \frac{\cos(\alpha^{n}) - 1}{\alpha^{m}} = \frac{-1}{2}$$

$$\lim_{\alpha \to 0} \frac{1 - \cos(\alpha^{n})}{\alpha^{2n}} \cdot \frac{\alpha^{2n}}{\alpha^{m}} = \frac{-1}{2}$$

$$\lim_{\alpha \to 0} \frac{1}{2} \cdot \alpha^{2n-m} = \frac{1}{2}$$

$$2n - m = 0 \Rightarrow \frac{m}{n} = 2$$
 Ans.

## **SECTION 2**

(Maximum Marks: 32)

- This section contains EIGHT questions
- Each question has FOUR options (A), (B), (C) and (D). ONE OR MORE THAN ONE of these four option(s) is(are) correct
- For each question, darken the bubble(s) corresponding to all the correct option(s) in the ORS
- Marking scheme:
  - +4 If only the bubble(s) corresponding to all the correct option(s) is (are) darkened
  - 0 If none of the bubbles is darkened
  - –2 In all other cases
- **49.** Let  $f(x) = 7 \tan^8 x + 7 \tan^6 x 3 \tan^4 x 3 \tan^2 x$  for all  $x \in \left(-\frac{\pi}{2}, \frac{\pi}{2}\right)$ . Then the correct expression(s) is(are)

(A) 
$$\int_{0}^{\pi/4} x f(x) dx = \frac{1}{12}$$

(B) 
$$\int_{0}^{\pi/4} f(x) dx = 0$$

(C) 
$$\int_{0}^{\pi/4} x f(x) dx = \frac{1}{6}$$

(D) 
$$\int_{0}^{\pi/4} f(x) dx = 1$$

Ans. AB

**Sol.** 
$$\sqrt{f(x)} = 7 \tan^6 x (\tan^2 x + 1) - 3 \tan^2 x (\tan^2 x + 1) = (7 \tan^6 x - 3 \tan^2 x) \sec^2 x$$

$$\int_{0}^{\pi/4} f(x) dx = \int_{0}^{\pi/4} (7 \tan^{6} x - 3 \tan^{2} x) \sec^{2} x dx$$

Put tan  $x = t \Rightarrow \sec^2 x dx = dt$ 

$$\int_{0}^{1} (7t^{6} - 3t^{2}) dt = 0$$

$$\int_{0}^{\pi/4} x f(x) dx = \int_{0}^{\pi/4} \underbrace{x}_{i} \underbrace{(7 \tan^{6} x - 3 \tan^{2} x)}_{ii} sec^{2} x dx$$

$$= \left(x \cdot (tan^7 \ x - tan^3)\right)_0^{\pi/4} - \int_0^{\pi/4} (tan^7 \ x - tan^3 \ x) dx$$

$$= 0 - \int_{0}^{\pi/4} \tan^{3} x(\tan^{4} x - 1) dx$$

$$= - \int_{0}^{\pi/4} \tan^{3} x(\tan^{2} x - 1) \sec^{2} x dx$$

tan x = t

$$sec^2x dx = dt$$

$$= -\int_{0}^{1} t^{3}(t^{2} - 1) dt = -\left(\frac{1}{6} - \frac{1}{4}\right) = \frac{1}{12}$$

Let  $f'(x) = \frac{192x^3}{2+\sin^4\pi x}$  for all  $x \in R$  with  $f\left(\frac{1}{2}\right) = 0$ . If  $m \le \int_{1/2}^1 f(x) dx \le M$ , then the possible values of m and 50.

M are

$$(A) m = 13, M = 24$$

(B) 
$$m = \frac{1}{4}$$
,  $M = \frac{1}{2}$   
(D)  $m = 1$ ,  $M = 12$ 

(C) 
$$m = -11$$
,  $M = 0$ 

(D) 
$$m = 1$$
,  $M = 12$ 

Ans.

**Sol.** 
$$f'(x) = \frac{192x^3}{2 + \sin^4 \pi x} \forall x \in R$$

Clearly f'(x) is increasing in  $\left| \frac{1}{2}, 1 \right|$ 

$$f'(x) \le f'(1) = 96$$

$$f'(x) \ge f'\left(\frac{1}{2}\right) = 8$$

for maximum,  $f(x) = y = 96\left(x - \frac{1}{2}\right)$ for minimum,  $f(x) = y = 96\left(x - \frac{1}{2}\right)$ 

$$\int_{1/2}^{1} f(x) dx \le \int_{1/2}^{1} 96 \left( x - \frac{1}{2} \right) dx = 12$$

$$= \int_{1/2}^{1} f(x) dx \ge \int_{1/2}^{1} 8 \left( x - \frac{1}{2} \right) dx = 4x^{2} - 4x \Big|_{1/2}^{1} = 0 - (1 - 2) = 1$$

$$\therefore 1 \leq \int_{1/2}^{1} f(x) dx \leq 12$$

- Let S be the set of all non-zero real numbers  $\alpha$  such that the quadratic equation  $\alpha x^2 x + \alpha = 0$  has two 51. distinct real roots  $x_1$  and  $x_2$  satisfying the inequality  $|x_1 - x_2| < 1$ . Which of the following intervals is(are) a subset(s) of S?
  - (A)  $\left(-\frac{1}{2}, -\frac{1}{\sqrt{5}}\right)$  (B)  $\left(-\frac{1}{\sqrt{5}}, 0\right)$  (C)  $\left(0, \frac{1}{\sqrt{5}}\right)$
- (D)  $\left(\frac{1}{\sqrt{5}}, \frac{1}{2}\right)$

Ans. AD

**Sol.** 
$$\alpha x^2 - x + \alpha = 0$$

$$1 - 4\alpha^2 > 0$$

$$4\alpha^2 - 1 < 0$$

$$\alpha \in \left(\frac{-1}{2}, \frac{1}{2}\right)$$

$$|x_1 - x_2| < 1$$

$$(x_1 - x_2)^2 < 1$$

$$(x_1 + x_2)^2 - 4x_1x_2 < 1$$

$$\left(\frac{1}{\alpha}\right)^2 - 4 < 1$$

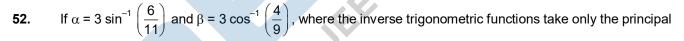
$$\frac{1}{\alpha^2} - 5 < 0$$

$$\frac{5\alpha^2-1}{\alpha^2}$$
 > 0

$$5\alpha^2 - 1 > 0, \alpha \neq 0$$

$$\therefore \alpha \in \left(\frac{-1}{\sqrt{5}}, \frac{1}{\sqrt{5}}\right) - \left\{0\right\}$$

$$\therefore \alpha \in \left(-\frac{1}{2}, \frac{-1}{\sqrt{5}}\right) \cup \left(\frac{1}{\sqrt{3}}, \frac{1}{2}\right)$$



values, then the correct option(s) is(are)

(A) 
$$\cos \beta > 0$$

(B) 
$$\sin \beta < 0$$

(C) 
$$\cos (\alpha + \beta) > 0$$

(D) 
$$\cos \alpha < 0$$

Ans. BCD

**Sol.** 
$$\alpha = 3 \sin^{-1} \left( \frac{6}{11} \right)$$

$$\sin^{-1}\left(\frac{1}{2}\right) < \sin^{-1}\left(\frac{6}{11}\right) < \sin^{-1}\left(\frac{1}{\sqrt{2}}\right)$$

$$3\left(\frac{\pi}{6}\right) < 3 \sin^{-1}\left(\frac{6}{11}\right) < 3\left(\frac{\pi}{4}\right)$$

$$\frac{\pi}{2} < \alpha < \frac{3\pi}{4}$$

 $\therefore \ \alpha \in \mathsf{II}^{\mathsf{nd}} \ \mathsf{quadrant}$ 

$$\beta = 3 \cos^{-1} \left( \frac{4}{9} \right)$$

$$\cos^{-1}\left(\frac{1}{2}\right) < \cos^{-1}\left(\frac{4}{9}\right) < \cos^{-1}\left(\frac{\sqrt{6} - \sqrt{2}}{8}\right)$$

$$3\left(\frac{\pi}{3}\right) < b < 3\frac{5\pi}{12}$$

$$\pi < \beta < \frac{5\pi}{4} \Rightarrow \beta \in \mathsf{III}^{\mathsf{rd}} \mathsf{quadrant}$$

Let  $E_1$  and  $E_2$  be two ellipses whose centers are at the origin. The major axes of  $E_1$  and  $E_2$  lie along the 53. x-axis and the y-axis, respectively. Let S be the circle  $x^2 + (y - 1)^2 = 2$ . The straight line x + y = 3 touches the curves S, E<sub>1</sub> and E<sub>2</sub> at P, Q and R, respectively. Suppose that PQ = PR =  $\frac{2\sqrt{2}}{3}$ . If e<sub>1</sub> and e<sub>2</sub> are the eccentricities of E<sub>1</sub> and E<sub>2</sub>, respectively, then the correct expression(s) is (are)

(A) 
$$e_1^2 + e_2^2 = \frac{43}{40}$$

(B) 
$$e_1 e_2 = \frac{\sqrt{7}}{2\sqrt{10}}$$

(B) 
$$e_1 e_2 = \frac{\sqrt{7}}{2\sqrt{10}}$$
 (C)  $\left| e_1^2 - e_2^2 \right| = \frac{5}{8}$ 

(D) 
$$e_1 e_2 = \frac{\sqrt{3}}{4}$$

Ans.

Sol.  $x - y + \lambda = 0$  normal to circle.

passes centre  $(0, 1) \Rightarrow \lambda = 1$ 

$$x - y + 1 = 0$$

$$x + y = 3$$

$$2x = 2$$

$$x = 1, y = 2$$

$$\frac{x-1}{\cos 135^{\circ}} = \frac{y-2}{\sin 135^{\circ}} = \pm \frac{2\sqrt{2}}{3}$$

$$x = 1 \pm x = 1 \pm \left(\frac{-2}{3}\right) = \frac{1}{3}, \frac{5}{3}$$

$$y = 2 \pm \left(\frac{2}{3}\right) = \frac{8}{3}, \frac{4}{3}$$

$$Q\left(\frac{5}{3}, \frac{4}{3}\right), R\left(\frac{1}{3}, \frac{8}{3}\right)$$

$$E_1: \frac{x^2}{a^2} + \frac{y^2}{b^2} = 1, (a > b)$$

$$T_Q: \frac{5x}{3a^2} + \frac{4}{3}\frac{y}{b^2} = 1$$

$$\frac{5}{3a^2} = \frac{4}{3b^2} = \frac{1}{3}$$

$$a^2 = 5$$
,  $b^2 = 4$ 

$$=e_1^2=1 - = \frac{b^2}{a^2} = \frac{1}{5}$$

$$E_2: \frac{x^2}{A^2} + \frac{y^2}{B^2} = 1$$
 (A < B)

$$T_R : \frac{1}{3A^2}x + \frac{8}{3B^2} = 1$$

Compare with x + y = 3

$$\frac{1}{3A^2} = \frac{8}{3B^2} = \frac{1}{3}$$

$$A^2 = 1$$

$$B^2 = 8$$

$$e_2^2 = 1 - \frac{1}{8} = \frac{7}{8}$$

$$e_1^2 + e_2^2 = \frac{8+35}{40} = \frac{43}{40}$$

$$\Rightarrow e_1 e_2 = \frac{\sqrt{7}}{2\sqrt{10}}$$

Consider the hyperbola  $H: x^2 - y^2 = 1$  and a circle S with center  $N(x_2, 0)$ . Suppose that H and S touch each other at a point  $P(x_1, y_1)$  with  $x_1 > 1$  and  $y_1 > 0$ . The common tangent to H and S at P intersects the x-axis at point M. If  $(\ell, m)$  is the centroid of the triangle  $\Delta PMN$ , then the correct expression(s) is(are)

(A) 
$$\frac{dI}{dx_1} = 1 - \frac{1}{3x_1^2}$$
 for  $x_1 > 1$ 

(B) 
$$\frac{dm}{dx_1} = \frac{x_1}{3(\sqrt{x_1^2 - 1})}$$
 for  $x_1 > 1$ 

(C) 
$$\frac{d\ell}{dx_1} = 1 + \frac{1}{3x_1^2}$$
 for  $x_1 > 1$ 

(D) 
$$\frac{dm}{dy_1} = \frac{1}{3}$$
 for  $y_1 > 0$ 

Sol ABD

**Sol.** 
$$H: x^2 - y^2 = 1$$

Tangent at P (x<sub>1</sub>, y<sub>1</sub>)

$$xx_1 - yy_1 = 1$$

$$M \equiv \left(\frac{1}{x_1}, 0\right)$$

Centroid of the triangle  $\Delta PMN$  is  $(\ell, m)$ 

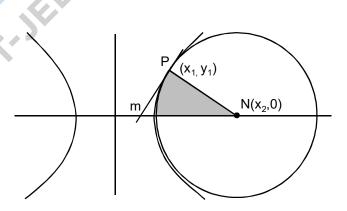
$$\ell = \frac{x_1 + \frac{1}{x_1} + x_2}{3}; m = \frac{y_1}{3}$$

Normal at P:  $\frac{x}{x_1} + \frac{y}{y_1} = 2$ , put y = 0

$$x = 2x_1 \Rightarrow x_2 = 2x_1$$

$$y_1^2 = x_1^2 - 1$$
.

$$\Rightarrow \ell = \frac{3x_1 + \frac{1}{x_1}}{3}$$



$$\frac{d\ell}{dx} = 1 - \frac{1}{3x^2} \Rightarrow (A)$$

$$m = \frac{y_1}{3}; m = \frac{\sqrt{x_1^2 - 1}}{3}$$

$$\frac{dm}{dy_1} = \frac{1}{3}; \frac{dm}{dy_1} = \frac{x_1}{3\sqrt{x_1^2 - 1}} \implies (B) \& (D)$$

**55.** The option(s) with the values of a and L that satisfy the following equation is (are)

$$\int\limits_{0}^{4\pi} e^{t} (\sin^{6} at + \cos^{4} at) dt \\ \int\limits_{0}^{\pi} e^{t} (\sin^{6} at + \cos^{4} at) dt$$

(A) 
$$a = 2$$
,  $L = \frac{e^{4\pi} - 1}{e^{\pi} - 1}$  (B)  $a = 2$ ,  $L = \frac{e^{4\pi} + 1}{e^{\pi} + 1}$  (C)  $a = 4$ ,  $L = \frac{e^{4\pi} - 1}{e^{\pi} - 1}$  (D)  $a = 4$ ,  $L = \frac{e^{4\pi} + 1}{e^{\pi} + 1}$ 

Ans. AC

$$\textbf{Sol.} \qquad I_{N} = \int\limits_{0}^{4\pi} e^{t} (sin^{6} \ at + cos^{4} \ at) dt$$

$$=\int\limits_0^\pi \underbrace{e^t(sin^6\ at+cos^4\ at)dt}_{\text{I}} + \int\limits_\pi^{2\pi} \underbrace{e^t(sin^6\ at+cos^4\ at)dt}_{\text{II}} + \int\limits_{2\pi}^{3\pi} \underbrace{e^{at}(sin^6\ at+cos^4\ at)dt}_{\text{III}} + \int\limits_{3\pi}^{4\pi} \underbrace{e^t(sin^6\ at+cos^4\ at)dt}_{\text{IV}}$$

Put  $t = \pi + v$ 

$$I_2 = \int_{0}^{\pi} e^{\pi} e^{v} (\sin^6 av + \cos^6 av) dv$$

$$= e^{\pi} \int_{0}^{\pi} e^{v} (\sin^{6} av + \cos^{4} av) dv$$

Put  $t = 2\pi + v$ 

$$I_3 = e^{2p} \int_{0}^{\pi} e^{v} (\sin^6 av + \cos^4 av) dv$$

Put  $t = 3\pi + v$ 

$$I_4 = e^{3p} \int_{0}^{\pi} e^{v} (\sin^6 av + \cos^4 av) dv$$

$$I_N = I_1 + I_2 + I_3 + I_4$$

= 
$$(1 + e^{\pi} + e^{2\pi} + e^{3\pi}) \int_{0}^{\pi} e^{t} (\sin^{6} at + \cos^{4} at) dt = \left(\frac{e^{4\pi} - 1}{e^{\pi} - 1}\right) I_{D}$$

$$\therefore \quad \frac{I_N}{I_D} = \frac{e^{4\pi} - 1}{e^{\pi} - 1}$$

**56.** Let f, g:  $[-1, 2] \rightarrow \mathbb{R}$  be continuous functions which are twice differentiable on the interval (-1, 2). Let the values of f and g at the points -1, 0 and 2 be as given in the following table:

	x = -1	x = 0	x = 2
f(x)	3	6	0
g(x)	0	1	-1

In each of the intervals (-1, 0) and (0, 2) the function (f - 3g)" never vanishes. Then the correct statement(s) is(are)

(A) f'(x) – 3g'(x) = 0 has exactly three solutions in (–1, 0)  $\cup$  (0, 2)

(B) f'(x) - 3g'(x) = 0 has exactly one solution in (-1, 0)

(C) f'(x) - 3g'(x) = 0 has exactly one solution in (0, 2)

(D) f'(x) - 3g'(x) = 0 has exactly two solutions in (-1, 0) and exactly two solutions in (0, 2)

Ans. BC

**Sol.** Consider h(x) = f(x) - 3g(x) on [-1, 0]

h(-1) = 3 = h(0) Rolle's Theorem

Applicable  $\Rightarrow \exists c \in (-1, 0)$  such that h'(c) = 0

 $\Rightarrow$  f '(c) - 3g'(c) = 0 for some c  $\in$  (-1, 0)

Clearly there exists only one  $x = c \in (-1, 0)$  for which h'(x) = 0.

because if  $\exists c_1 \neq c$  for which h'(x) = 0 then again by Rolle's Theorem

h''(x) = 0 for some  $x \in (-1, 0)$ 

i.e. f''(x) - 3g''(x) = 0 for some  $x \in (-1, 0)$  which is not possible  $\Rightarrow$  (B)

Again h(0) = 3, h(2) = 3

again by same argument f'(x) - 3g'(x) = 0 for some x in (0, 2) and such x is unique.  $\Rightarrow$  (C)

#### **SECTION 3**

(Maximum Marks: 16)

- . This section contains TWO paragraphs
- . Based on each paragraph, there will be TWO questions
- . Each question has FOUR options (A), (B), (C) and (D). One or more than one of these four option(s) is(are) correct
- . For each question darken the bubble(s) corresponding to all the correct option(s) is the ORS
- . Marking scheme:
  - + 4 If only the bubble (s) corresponding to all the correct option(s) is/are darkened
  - 0 If none of the bubbles is darkened
  - 2 In all other cases

#### PARAGRAPH - 1

Let n<sub>1</sub> and n<sub>2</sub> be the number of red and black balls, respectively, in box I. Let n<sub>3</sub> and n<sub>4</sub> be the number of red and black balls, respectively, in box II.

57. One of the two boxes, box I and box II, was selected at random and a ball was drawn randomly out of this box. the ball was found to be red. If the probability that this red ball was drawn from box II is  $\frac{1}{2}$ , then the correct option(s) with the possible values of n<sub>1</sub>, n<sub>2</sub>, n<sub>3</sub> and n<sub>4</sub> is(are)

(A) 
$$n_1 = 3$$
,  $n_2 = 3$ ,  $n_3 = 5$ ,  $n_4 = 15$ 

(B) 
$$n_1 = 3$$
,  $n_2 = 6$ ,  $n_3 = 10$ ,  $n_4 = 50$ 

(C) 
$$n_1 = 8$$
,  $n_2 = 6$ ,  $n_3 = 5$ ,  $n_4 = 20$ 

(D) 
$$n_1 = 6$$
,  $n_2 = 12$ ,  $n_3 = 5$ ,  $n_4 = 20$ 

Ans. A, B

58. A ball is drawn at random from box I and transferred to box II. If the probability of drawing a red ball from box I, after this transfer, is  $\frac{1}{2}$ , then the correct option(s) with the possible values of  $n_1$  and  $n_2$  is(are)

(A) 
$$n_1 = 4$$
 and  $n_2 = 6$ 

(B) 
$$n_1 = 2$$
 and  $n_2 = 3$ 

$$(C^*)$$
  $n_1 = 10$  and  $n_2 = 20$ 

$$(D^*) n_1 = 3 \text{ and } n_2 = 6$$

Sol.

$$n_1R$$
  $n_2B$   $n_3R$   $n_4E$ 

(C\*) 
$$n_1 = 10$$
 and  $n_2 = 20$  (D\*)  $n_1 = 3$  and  $n_2 = 6$ 

Sol.

I

 $n_1R$ 
 $n_2B$ 
 $n_3R$ 
 $n_4B$ 

(i)

Probability

$$\frac{\frac{1}{2} \times \frac{n_3}{n_3 + n_4}}{\frac{1}{2} \times \frac{n_1}{n_1 + n_2} + \frac{1}{2} \times \frac{n_3}{n_3 + n_4}} = \frac{n_3 (n_1 + n_2)}{n_1 (n_3 + n_4) + n_3 (n_1 + n_2)} = \frac{1}{3}$$

(A)

 $n_1 = 3, n_2 = 3, n_3 = 5, n_4 = 15$ 

$$\therefore \text{L.H.S.} = \frac{5 \cdot 6}{3 \cdot 20 + 5 \cdot 6} = \frac{30}{90} = \frac{1}{3}.$$

(A) 
$$n_1 = 3, n_2 = 3, n_3 = 5, n_4 = 15$$

$$\therefore$$
 L.H.S.  $=\frac{5\cdot 6}{3\cdot 20+5\cdot 6}=\frac{30}{90}=\frac{1}{3}$ .

Hence, (A) is correct

**(B)** 
$$n_1 = 3$$
,  $n_2 = 6$ ,  $n_3 = 10$ ,  $n_4 = 50$ 

$$\therefore$$
 L.H.S.  $=\frac{10\times9}{3\times60+10\times9}=\frac{90}{270}=\frac{1}{3}$ 

Hence, (B) is correct

(C) 
$$n_1 = 8$$
,  $n_2 = 6$ ,  $n_3 = 5$ ,  $n_4 = 20$ 

$$\therefore$$
 L.H.S.  $=\frac{5\times14}{8\times25+5\times14}=\frac{70}{270}\neq\frac{1}{3}$ 

**(D)** 
$$n_1 = 6$$
,  $n_2 = 12$ ,  $n_3 = 5$ ,  $n_4 = 20$ 

$$\therefore \text{ L.H.S.} = \frac{5 \times 18}{6 \times 25 + 5 \times 18} = \frac{90}{240} \neq \frac{1}{3}.$$

:. Answer is (A) & (B)

(ii) Probability = 
$$\frac{n_1}{n_1 + n_2} \times \frac{(n_1 - 1)}{(n_1 + n_2 - 1)} + \frac{n_2}{n_1 + n_2} \times \frac{n_1}{(n_1 + n_2 - 1)} = \frac{1}{3}$$

(A) 
$$n_1 = 4$$
,  $n_2 = 6$ 

L.H.S. 
$$=\frac{4}{10} \times \frac{3}{9} + \frac{6}{10} \times \frac{4}{9} = \frac{36}{90} \neq \frac{1}{3}$$

**(B)** 
$$n_1 = 2, n_2 = 3$$

L.H.S. 
$$=\frac{2}{5} \times \frac{1}{4} + \frac{3}{5} \times \frac{2}{4} = \frac{8}{20} \neq \frac{1}{3}$$

(C) 
$$n_1 = 10, n_2 = 20$$

L.H.S. 
$$=\frac{10}{30} \times \frac{9}{29} + \frac{20}{30} \times \frac{10}{29} = \frac{290}{870} = \frac{1}{3}$$

Hence, (C) is correct

**(D)** 
$$n_1 = 3, n_2 = 6$$

L.H.S. 
$$=\frac{3}{9} \times \frac{2}{8} + \frac{6}{9} \times \frac{3}{8} = \frac{24}{72} = \frac{1}{3}$$

Hence, (D) is correct.

:. Answer is (C) & (D).

#### PARAGRAPH - 2

Let F:  $\mathbf{R} \to \mathbf{R}$  be a thrice differentiable function. Suppose that F(1) = 0, F(3) = -4 and F'(x) < 0 for all

$$x \in \left(\frac{1}{2}, 3\right)$$
. Let  $f(x) \equiv x F(x)$  for all  $x \in R$ .

(B) 
$$f(2) < 0$$

(C) f'(x) 
$$\neq$$
 0 for any x  $\in$  (1,3)

(D) f'(x) = 0 for some 
$$x \in (1,3)$$

Ans. ABC

**60.** If 
$$\int_{1}^{3} x^2 F'(x) dx = -12$$
 and  $\int_{1}^{3} x^3 F''(x) dx = 40$  then the correct expression(s) is(are)

(A) 
$$9f'(3) + f'(1) - 32 = 0$$

(B) 
$$\int_{0}^{3} f(x) dx = 12$$

(C) 
$$9f'(3) - f'(1) + 32 = 0$$

(D) 
$$\int_{1}^{3} f(x) dx = -12$$

Ans. CD

Sol.

(i) 
$$\underbrace{F'(x) < 0}_{\text{decreasing function}}, \ x \in \left(\frac{1}{2}, 3\right)$$

$$f(x) = x F(x)$$

$$F(1) = 0, F(3) = -4$$

$$f'(x) = x F'(x) + F(x)$$

$$f'(1) = F'(1) + F(1) = F'(1) < 0 \Rightarrow (A)$$

$$f(2) = 2F(2)$$

Since, F is decreasing and F (1) = 0  $\Rightarrow$  F (2) < 0

$$\therefore \qquad f(2) = 2 F(2) < 0 \qquad \Rightarrow$$

Again f'(x) = 
$$\underbrace{x F'(x)}_{0} + \underbrace{F(x)}_{0} \forall x \in (1, 3)$$

$$\Rightarrow$$
 f'(x) < 0  $\forall$  x  $\in$  (1, 3)  $\Rightarrow$  f'(x)  $\neq$  0  $\forall$  x  $\in$  (1, 3)  $\therefore$  Ans. (C)

(B)

COUND ATIO

## (ii) [CD]

$$\int_{1}^{3} x^{2} F'(x) dx = -12; \int_{1}^{3} x^{3} F''(x) dx = 40$$

$$\int_{1}^{3} x^{2} F'(x) dx = x^{2} F(x) \Big|_{1}^{3} - 2 \int_{1}^{3} x F(x) dx = -12$$

= 
$$-36 - 2 \int_{3}^{3} f(x) dx = -12 \Rightarrow \int_{3}^{3} f(x) dx = -12 \Rightarrow (D)$$

Now, 
$$\int_{1}^{3} x^{3} F''(x) dx = x^{3} F'(x) \Big|_{1}^{3} - 3 \int_{1}^{3} x^{2} F'(x) dx$$

= 27 F'(3) - F'(1) - 3 
$$\left[ x^2 F(x) \right]_1^3 - 2 \int_1^3 x F(x) dx$$

= 27 F'(3) - F'(1) - 3 
$$\left[ 9F(3) - F(1) - 2 \int_{3}^{3} f(x) dx \right]$$

$$= 27F'(3) - 27F(3) - F'(1) + 3F(1) + 6 \times (-12)$$

$$= 27F'(3) + 108 - F'(1) - 72 = 9 (3 F'(3)) - F'(1) + 36$$

$$= 9 (f'(3) - F(3)) - (f'(1) - F(1)) + 36$$

$$= 9 f'(3) + 36 - f'(1) + 36 = 40$$

$$\Rightarrow$$
 9f '(3) – f '(1) + 32 = 0  $\Rightarrow$  (C)]